Revision: 000 Date: 19 May 2019

1. IDENTIFICATION OF THE SUBSTANCE AND COMPANY UNDERTAKING

General Chemical Name MOLYBDENUM NITRIDE powder

Intended/recommended use: Research

Supplier (Distributor): Heeger Materials Inc.

230 Steele St Denver,

CO 80206 United States Tel: 925-385-8104

2. COMPOSITION/INFORMATION ON INGREDIENTS

Ingredient Name Concentration Classification* CAS number EINECS No.

Molybdenum nitride 100% Xn: R48/20/22 12033-01-1

Xi: R36/37/38

*see 16. OTHER INFORMATION for full text of R-phrases.

3. HAZARD IDENTIFICATION

Danger of serious damage to health by prolonged exposure through inhalation and if swallowed.

Irritating to eyes, respiratory system and skin.

4. FIRST AID MEASURES

Inhalation If signs/symptoms like coughing or burning occur, remove person from

exposure to fresh air immediately and administer 100 percent humidified supplement oxygen with assisted ventilation as required. If breathing has ceased DO NOT use mouth-to-mouth respiration. Apply artificial respiration using oxygen and a suitable mechanical device such

as a bag and a mask. Seek immediate medical attention.

Skin Contact Flush skin with large amounts of water. Remove contaminated clothing.

If irritation persists, seek medical attention

Eye Contact DO NOT allow patient to rub or keep eyes closed. Irrigate with copious

quantities of water for at least 15 minutes. Flush under eyelids by lifting lid. DO NOT use a static eye bath. Seek immediate medical attention.

Induce vomiting if patient is conscious and alert. Give 2 cupfuls of milk

with great care. Do not induce vomiting and give nothing by mouth if

patient is unconscious. Seek immediate medical attention.

5. FIRE FIGHTING MEASURES

Precautions against fire and explosion

Material is non flammable but will react with water or steam to produce flammable, toxic and corrosive gas (ammonia).

Use dry sand, dry salt, carbon dioxide or extinguishing powder. For large fires cool containers with flooding quantities of water until well after the fire is out. Do not get water inside containers.

Extinguishing media which must not be used for safety reasons

DO NOT USE water or foam, See above.

Exposure hazards arising from substance, combustion products, resulting gases

Molybdenum nitride will react with water to produce flammable, toxic and corrosive gas (ammonia).

Molybdenum oxide particles may be formed in fire which are toxic and irritant.

Nitrogen oxides may also be formed.

Special protective equipment for fire fighters

Wear full protective clothing, including self-contained breathing apparatus.

6. ACCIDENTAL RELEASE MEASURES

Personal Precautions:

Remove or extinguish all ignition sources. Evacuate all but essential authorised control personnel. Wear self-contained breathing apparatus and gloves to avoid inhalation, skin and eye contact. Provide sufficient ventilation.

Environmental Precautions

Prevent entry into drains, surface and ground water, soil and confined areas.

Methods for Cleaning up

Only trained, authorised personnel should be involved. Fully encapsulating protective clothing and self contained breathing apparatus should be worn for spills and leaks with no fire. Do not touch damaged containers or spilled material unless wearing appropriate protective clothing.

Stop leak if you can do it without risk. Keep combustibles (wood, paper, oil etc) away from spilled material. Cover spills with dry sand, dry diatomaceous earth or dry salt followed with plastic sheet to minimise spreading or contact with water. Use clean nonsparking tools to collect material and place it into loosely covered plastic containers, suitably marked, and dispose of through a licensed disposal contractor.

7. HANDLING AND STORAGE

Precautions for safe handling

To be handled by qualified and trained staff only. Avoid breathing dusts and direct contact with skin and eyes. Avoid all contact with water. Wash hands thoroughly after handling. See section 8 for personal protective equipment.

Handle under dry inert gas. Ensure good ventilation/exhaustion at the workplace

Keep ignition sources away - do not smoke
Protect against electrostatic charges.
Fumes can combine with air to form an explosive mixture.

Precautions for safe storage

Keep container cool, dry and tightly closed when not in use. Moisture sensitive. Store away from oxidisers and other materials listed under incompatibility (see section 10).

Specific use

Research.

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Exposure limit values

Ingredient name OES LTEL OES STEL MEL(LT) MEL(ST)

(8 hr TWA) (15 min)

Molybdenum compounds

(as Mo) insoluble compounds 10mg/m^3 20mg/m^3

The usual precautionary measures should be adhered to in handling chemicals. Keep away from foodstuffs, beverages and food. Instantly remove any soiled and impregnated garments. Wash hands during breaks and at end of the work. Avoid contact with the eyes and skin.

Occupational exposure controls

Eye Protection Wear appropriate protective eyeglasses, chemical safety goggles

or full face shield to European Standard EN 166.

Hand protection Wear appropriate gloves when handling this material. Suitable

material is butyl rubber. However, due regard must be taken that heavy gloves will interfere with the wearer's sense to touch and may contribute to a dangerous situation. Thinner gloves of nitrile

and PVC may be used as disposable gloves and must be discarded immediately after use. Gloves should comply with

European Standard EN 465-3 class 3.

Skin Protection Use one or more of the following personal protection items as

necessary to prevent skin contact: Full chemical protective suit

to EN 465 standard, PVC apron, helmet and boots.

Respiratory Protection Avoid inhalation of dust. Select the following respirator based

on airborne concentration of contaminants: Full face dust respirator. Half -mask air-supplied respirator to EN 147. Full-

face high efficiency filter respirator to EN 147 or EN 12941/12942. Full-face or hood compressed air breathing apparatus to EN 139 or EN 270/271. Use appropriate local

exhaust ventilation, to maintain airborne exposure below control

limits.

Ingestion: Do not eat, drink or smoke when using this product. Do not

ingest. Exhibit the strictest hygiene control.

Environmental exposure controls

No specific environmental legislation applies, however in accordance with best practise only use in a fume cupboard or with local exhaust ventilation, ventilated to a scrubbing system. In case of fire, spillage, or leakage, prevent material from entering water courses, sewers or soil. Material will decompose in air forming ammonia gas.

9. PHYSICAL AND CHEMICAL PROPERTIES

General information

Appearance Black powder

Odour Slight ammoniacal odour

Important health, safety and environmental information

pН N/A

Boiling point/boiling range

Melting point/melting range

Flash point

N/A
N/A
Crials
n/c Flammability (solid, gas)

Autoignition temperature

Explosive properties N/A

Oxidising properties Not oxidising

Vapour Pressure N/A

9.06g/cm³ at 20⁰C Relative density

Solubility

Water Reacts

10. STABILITY AND REACTIVITY

Stable, but will decompose if exposed to moist air or

water.

Conditions to avoid Avoid creating dusts. Air and moisture sensitive.

Materials to avoid Avoid water, acids, oxidising agents.

Hazardous Decomposition Products Will react with water at room temperature to

produce flammable/explosive and corrosive ammonia gas. Decomposition products include nitrogen oxides

and molybdenum oxides.

11. TOXICOLOGICAL INFORMATION

Molybdenum nitride will react with water under normal conditions.

Effects from eye contact May cause severe irritation and damage to the eyes.

Symptoms include burns, irritation, redness and conjunctivitis. Repeated or prolonged exposure may cause irreversible damage to the conjunctiva, cornea and

lens.

Effects from skin contact May cause skin irritation and burns. Prolonged exposure

may cause tissue damage.

Effects from inhalation May cause irritation to mucous membranes and upper

respiratory tract. Ammonia gas may cause irritation to the nose and throat, dyspnea, bronchial spasms, chest pains, pulmonary edema and pink frothy sputum.

Effects from ingestion Irritating to digestive tract. May cause acute or

chronic molybdenum poisoning. Will decompose to toxic and corrosive ammonia causing secondary damage. Symptoms include abdominal pain, nausea, vomiting soreness/redness of the mouth and throat, burns and

dysphagia.

(a) Acute toxicity (oral, inhalation, dermal) not tested/no data

(b) Corrosive/irritation (eye, skin, respiratory tract) not tested/no data

(c) Sensitisation (skin, respiratory) not tested/no data

(d) Repeated-dose toxicity not tested/no data

(e) Mutagenicity not tested/no data

(f) Carcinogenicity not tested/no data

(g) Reproductive toxicity not tested/no data

To the best of our knowledge the chemical, physical and toxicological properties of molybdenum nitride have not been thoroughly investigated and recorded.

Molybdenum compounds are toxic by subcutaneous and intraperitoneal routes. Molybenum and its compounds are highly toxic based upon animal experiments. Symptoms of acute poisoning include severe gastrointestinal irritation with diarrhoea, coma and death from heart failure. Experimental animals exposed to high levels accumulated molybdenum in the lungs, spleen and heart and showed a decrease of DNA and RNA in the liver, kidneys and spleen

Many nitrides react with moisture to evolve ammonia which is a human poison by an unspecified route.

Mate

12. ECOLOGICAL INFORMATION

(1) <u>Ecotoxicity</u>

- (a) Aquatic toxicity
 - (i) acute and chronic for fish not tested/no data
 - (ii) acute and chronic for daphnia not tested/no data
 - (iii) acute and chronic for algaefish not tested/no data
 - (iv) acute and chronic for other aquatic plants not tested/no data
- (b) Soil toxicity

(i)	macro organisms	not tested/no data
(ii)	micro organisms	not tested/no data
(iii)	birds	not tested/no data
(iv)	bees	not tested/no data
(v)	plants	not tested/no data
(vi)	fauna	not tested/no data

(2) Mobility

The product's mobility is limited because it will react with water or air-moisture at room temperature.

(3) Persistence and degradability

The product will rapidly react with water or moisture and degrade.

(4) Bioaccumulation potential

Molybdenum nitride cannot enter the food chain because it will immediately react with water and form toxic molybdenum oxide.

Other adverse effects (5)

No other effects are known.

13. DISPOSAL CONSIDERATIONS

All waste material to be contained in a plastic sealed bin, duly marked, and disposed of as special waste through a licensed waste contractor in accordance with "The Special Waste Regulations 1996". See sections 6 and 7.

14. TRANSPORT INFORMATION

- (a) UN number
- (b) class
- proper shipping name: (c)
- 1 2 3288 6.1c Toxic solid, inorganic, n.o.s. (molybdenum als Inc.
 - nitride _

- packing group (d)
- marine pollutant (if applicable) (e)
- other applicable information (f)
- IIIN/A
- N/A

15. REGULATORY INFORMATION

EC Supply: Chip-3 regulations 2002

> Harmful Xn: Xi: **Irritant**

Risk Phrases:

48/20/22: Harmful: danger of serious damage to health by prolonged exposure through

inhalation and if swallowed.

36/37/38: Irritating to eyes, respiratory system and skin.

Safety Phrases:

7/8: Keep container tightly closed and dry.

Do not breathe dust. 22:

In case of contact with eyes, rinse immediately with plenty of water and seek 26: medical advice

This material is subject to the COSHH regulations 2002. See COSHH Essentials for further information.

16. OTHER INFORMATION

R Phrases Full Text:

48/20/22: Harmful: danger of serious damage to health by prolonged exposure

through inhalation and if swallowed.

36/37/38: Irritating to eyes, respiratory system and skin

Abbreviations used:

Chemical Abstracts Service Registry Numbers **CAS**

European Inventory of Existing Commercial Chemical Substances **EINECS**

Material Safety Data Sheet **MSDS**

HSE Health and Safety Executive

TWA Time Weighted Average

OES Occupational Exposure Standards

Tials I This material should only be handled by qualified, trained chemists, fully familiar with its dangerous properties. During use or handling, a minimum of two persons should always be available.

References:

Chemicals (Hazard Information and Packaging for Supply) Regulations 2002

Approved classification and labelling guide. Chemicals (Hazard Information and Packaging for Supply) Regulations 2002 Guidance on Regulations L131

Approved Supply List. Information Approved for the Classification and labelling of Substances and Preparations Dangerous for Supply. Chemicals (Hazard Information and Packaging for Supply) Regulations 2002. Approved *List L129*

Control of Substances Hazardous to Health Regulations 2002

Health and Safety at Work Act 1974

COSHH Essentials: Easy Steps to Control Chemicals. Control of Substances Hazardous to Health Regulations

Occupational Exposure Limits 2001/2002 EH40

First Aid at Work. The Health and Safety (First Aid) Regulations 1981. Approved Code of Practice and Guidance L74

Personal Protective Equipment (EC Directive) Regulations 1992

The Selection, Use and Maintenance of Respiratory Protective Equipment: A Practical Guide HSG53

Cost and Effectiveness of Chemical Protective Gloves for the Workplace.

Guidance for Employers and Health and Safety Specialists. HSG206

Environmental Protection Act 1990 c43

Environmental Act 1995 c25

The Special Waste Regulations 1996

The Dangerous Substances and Explosive Atmospheres Regulations 2002

Inc.